Effective One-pot Synthesis of H type 1 and 2 Trisaccharide Derivatives Using Glycal Epoxide

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(Received December 21, 2004; CL-041577)

We describe an efficient synthesis of H type 1 and 2 trisaccharides by one-pot glycosylation involving glycosidation of glycal epoxide.

Oligosaccharides play important roles in biological processes on cell surface and have served as important tumor markers. β -Galactoside **1** attached with α -fucoside at the C2 position (H-disaccharide) is often found in biologically active oligosaccharides such as H type 1 and 2 epitopes (**1a**) and (**1b**), and is known to be an appropriate tumor antigen (Scheme 1).¹ In order to develop chemical probes based on the structure of the H-disaccharide, an effective methodology for the synthesis of glycoconjugates containing the H-disaccharide is required.²





One-pot sequential glycosylation to form two and more glycosidic bonds, is an effective approach for the liquid-phase oligosaccharide synthesis.³ This approach involves sequential chemo- and regio-selective glycosylations without any protecting group manipulations and purification of each intermediate. In the one-pot glycosylation, reagents and resulting products should not interfere any following glycosylations. We have investigated one-pot glycosylation based on the chemoselective activation of various glycosyl donors with an appropriate activator, and recently reported the synthesis of a protected linear and branched trisaccharide libraries by the one-pot glycosylation method.⁴ Most of synthetic strategies are based on the in situ synthesis of oligosaccharides with a leaving group. Therefore, the synthesis of the trisaccharides 1 based on the one-pot glycosylation strategy requires the glycosidation of glycosyl donor attached with saccharide at the C2 position to from 1,2-trans-glycosidic bonds. However, the absence of the participating substituents of the glycosyl donors at the C2 position makes it difficult to stereoselectively form the 1,2-trans-glycosidic bond. Therefore, an effective one-pot glycosylation method for the synthesis of various C2 glycosylated oligosaccharides is required. Herein we report the one-pot synthesis of H Type 1 and 2 trisaccharide units using a glycal epoxide.

Our strategy for the one-pot synthesis of H Type 1 and 2 trisaccharide units 1 is based on the preparing glycosyl acceptors 5 by glycosylation of acceptor 3 with glycal epoxide 2. The glycal epoxides are known to undergo stereoselective glycosidation to provide glycosides 5 bearing a hydroxy group at the C2 position linked through a 1,2-*trans*-glycosidic bond.⁵ Subsequent glycosylation of the hydroxy group with glycosyl donor 4 would provide the protected H Type 1 and 2 trisaccharides 1.

We first conducted the one-pot synthesis of H type 1 trisaccharide **9aA** bearing a primary amino group using the three building blocks **2**, **3a**, and **4A** (Scheme 2). Our initial investigation involves the stepwise synthesis of the protected trisaccharide **7aA** as shown in Scheme 2. Treatment of glucosamine **3a** with 1.2 equiv. of the glycal epoxide **2** in the presence of ZnCl₂ in CH₂Cl₂ provided the desired disaccharide **6** in 74% yield with complete β -selectivity. Use of TMSOTf as an activator resulted in a significant amount of silylated products. Fucosylation of the resulting disaccharide **6** with thiofucoside **4A** smoothly proceeded in stereoselective manner to provide α -fucoside **7aA** in good yield with complete α -selectivity.



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Chemistry Letters Vol.34, No.3 (2005)

Next, we examined one-pot glycosylation using 2, 3a, and 4A. To glucosamine 3a was added 1.2 equiv. of the glycal epoxide 2 and 2.5 equiv. of $ZnCl_2$ at -78 °C. The reaction mixture was stirred at 0 °C for 2h. Subsequently, 2.5 equiv. of thiofucoside 4A, 3.0 equiv. of NIS, and a catalytic amount of TfOH at 0 °C were added to the reaction mixture. After stirring at the same temperature for 2h, the reaction mixture was quenched. After removal of the solvent, the residue was purified by silica gel chromatography and gel permeable chromatography to provide trisaccharide 7aA in 46% yield based on 3a. The analytical data of trisaccharide 7aA, synthesized by one-pot glycosylation were identical with those of trisaccharide 7aA by the stepwise synthesis.

Deprotection of the protected trisaccharide **7aA** was investigated. Treatment of **7aA** with $NH_2NH_2 \cdot H_2O$ in EtOH at reflux for 18 h, followed by acetylation of the amine provided acetamide **8aA** in 70% yield. Hydrogenolysis of both benzyl ethers and an azido group with H_2 in the presence of Pd/C provided the H type 1 trisaccharide **9aA** bearing with an amino alkyl chain at the reducing end in quantitative yield.

In order to demonstrate the feasibility of the method, we planned the combinatorial synthesis of a small oligosaccharide library **7aA-cB** based on the structure of H type 1 and 2 trisaccharide by one-pot glycosylation. (Figure 1 and Scheme 3). Six building blocks 2, **3a–3c**, and **4A–4B** were designed for



Scheme 3. Reagents and conditions: a) $ZnCl_2$, CH_2Cl_2 , -78 to -20 °C, 2.0 h; b) NIS, TfOH, CH_2Cl_2 , 0 °C, 2.0 h.

the library synthesis.

The parallel synthesis of the 6 oligosaccharides 7 by one-pot glycosylation was performed utilizing Carusel[®], which controls the reaction temperature and the stirring rate in 10 reaction vessels. The six reaction vessels were set up with activated MS-4Å. Each acceptor 3a-3c was added to the two reaction vessels, respectively and the reaction vessels were cooled to -78 °C. The glycal epoxide 2 (1.2 equiv.) and ZnCl₂ (2.5 equiv.) were added to all the vessels at -78 °C. The reaction mixtures were warmed to -20 °C and stirred for 2 h at the same temperature. Subsequently, 2.5 equiv. of thiofucoside 4A (2.5 equiv.) or thiorthamnoside 4B (2.5 equiv.), NIS (3.0 equiv.), and a catalytic amount of TfOH at 0 °C were added to the reaction mixture. After stirring for 2h at the same temperature, the reaction mixture was quenched with NEt₃. The residues were purified by silica gel chromatography, followed by gel permeable chromatography to provide trisaccharide 7aA-bC in moderate yields (42-60% yields) based on 3.

In conclusion, we have demonstrated one-pot synthesis of H type 1 and 2 trisaccharides **7** using the glycal epoxide **2**. The glycosydation of glycal epoxide **2** with $ZnCl_2$ provided disaccharide possessing the hydroxy-free C2. The secondary hydroxy group subsequent undergo glycosylation to provide trisaccharide in good yield in one-pot. The one-pot sequential glycosylation should be useful to prepare oligosaccharides containing the H-disaccharide moiety.

References

- a) S. Hakomori and Y. Zhang, *Chem. Biol.*, 4, 97 (1997).
 b) D. S. Newburg, *Curr. Med. Chem.*, 6, 117 (1999).
- 2 a) S. J. Danishefsky, V. Behar, J. T. Randolph, and K. O. Lloyd, *J. Am. Chem. Soc.*, **117**, 5701 (1995). b) K. R. Love, R. B. Andrade, and P. H. Seeberger, *J. Org. Chem.*, **66**, 8165 (2001).
- 3 a) S. Raghavan and D. Kahne, J. Am. Soc. Chem., 115, 1580 (1993). b) S. V. Ley and H. W. M. Pripke, Angew. Chem., Int. Ed., 33, 2292 (1994). c) H. K. Chenault and A. Castro, Tetrahedron Lett., 35, 9145 (1994). d) T. Tsukida, M. Yoshida, K. Kurosawa, Y. Nakai, T. Achiha, T. Kiyoi, and H. Kondo, J. Org. Chem., 62, 6876 (1997). e) X.-S. Ye and C.-H. Wong, J. Org. Chem., 65, 2410 (2000).
- a) H. Yamada, T. Harada, and T. Takahashi, J. Am. Chem. Soc., 116, 7919 (1994). b) H. Yamada, T. Harada, T. Miyazaki, and T. Takahashi, Tetrahedron Lett., 35, 3979 (1994). c) H. Yamada, T. Kato, and T. Takahashi, Tetrahedron Lett., 40, 4581 (1999). d) T. Takahashi, M. Adachi, A. Matsuda, and T. Doi, Tetrahedron Lett., 41, 2599 (2000). e) H. Yamada, H. Takimoto, T. Ikeda, H. Tsukamoto, T. Harada, and T. Takahashi, Synlett, 2001, 1751. f) H. Tanaka, M. Adachi, H. Tsukamoto, T. Ikeda, H. Yamada, and T. Takahashi, Org. Lett., 4, 4213 (2002). g) H. Tanaka, M. Adachi, and T. Takahashi, Tetrahedron Lett., 45, 1433 (2004). h) M. Adachi, H. Tanaka, and T. Takahashi, Synlett, 2004, 609. i) H. Tanaka, M. Adachi, and T. Takahashi, Chem.—Eur. J., (2005), in press.
- 5 a) J. T. Randolph and S. J. Danishefsky, Angew. Chem., Int. Ed., 33, 1470 (1994). b) M. T. Bilodeau, T. K. Park, S. Hu, J. T. Randolph, S. J. Danishefsky, P. O. Livingston, and S. Zhang, J. Am. Chem. Soc., 114, 7840 (1995). c) S. J. Danishefsky and M. T. Biodeau, Angew. Chem., Int. Ed., 35, 1380 (1996). d) D. Sames, X.-T. Chen, and S. J. Danishefsky, Nature, 389, 587 (1997). e) J. T. Randolph, K. F. McClure, and S. J. Danishefsky, J. Am. Chem. Soc., 117, 5712 (1995).